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Reset Utility v 2.22. Serial Number. Study of polarizability and electric conductivity for liquid water in the hydrophobic cavity of a zeolite. We have measured the surface and bulk electric conductivity and the frequencydependent dielectric function of water confined in the hydrophobic cavity of the zeolite Y, at ambient temperature (298 K) using a cavity model and a lowfrequency impedance spectroscopic technique. The main finding in this work is that the conductivity of the confined liquid water is considerably

larger than that of free water. This is a direct evidence of the confinement effect in the zeolite Y, which results from the full filling of the hydrophobic cavity by liquid water. We also measured the frequency-dependent dielectric function of water at several levels of confinement. We reveal that the contribution of the permittivity attributed to the water-water interaction to the total dielectric function is smaller than that of water in the hydrophobic cavity of zeolite Y. In addition, we confirmed that the

significant difference between the dielectric response of water in the hydrophobic cavity of zeolite Y and that of bulk water is mainly due to the confinement-induced shifts in the frequency of the so-called "indirect" and "direct" water dipolar relaxation processes. The electric conductivity and dielectric function of confined water in the cavity of zeolite Y reported in this paper are for the first time compared with those for the confined water and liquid water in a model hydrophobic host, C18-6, reported in

our preceding papers [M. Kamata, Y. Nakai, K. Tsuda, and T. Suzuki, and Y. Nakai, K. Tsuda, and T. Suzuki, Phys. Rev. E 75, 031803 (2007); and 79a2804d6b

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